



# **ELECTRICITY AND ITS PRODUCTION**

# WHAT IS ELECTRICITY ?

- Electricity – electrical energy and movement of charge
- Electrical Power (P) – the rate of transformation of electrical energy.
  - Measured in watts
  - $1 \text{ W} = 1 \text{ J/s}$

$$P = \frac{\Delta E}{\Delta t}$$



$$1 \text{ W} = \frac{1 \text{ J}}{\text{s}}$$

## PRACTICE PROBLEMS ON PG. 505

### Practice

1. What is the power rating of a digital camera that transforms 120 J in 25 s? T/I  
[ans: 4.8 W]
2. How much power does a hair dryer require to transform 198 000 J of energy in 15 min? T/I [ans: 220 W]

$$\begin{aligned}
 1. \quad P &= \frac{\Delta E}{\Delta t} \\
 P &= \frac{120 \text{ J}}{25 \text{ s}} \\
 P &= 4.8 \text{ W}
 \end{aligned}$$

$$2. \quad 15 \cancel{\text{ min}} \times \frac{60 \text{ s}}{\cancel{\text{ min}}} = 900 \text{ s}$$

$$\begin{aligned}
 P &= \frac{\Delta E}{\Delta t} \\
 P &= \frac{198\,000 \text{ J}}{900 \text{ s}} \\
 P &= 220 \text{ W}
 \end{aligned}$$

# MEASURING ELECTRICAL ENERGY

- Measured in kilowatt hours (kWh)
- kilowatt hour (kWh) – measure of electrical energy
- $1 \text{ kWh} = 3.6 \text{ million joules}$
- To describe electrical energy generated by a power plant use megawatt hours (MWh)



# PRACTICE PROBLEMS PG. 506

## Practice

1. A compact fluorescent light bulb operates with 7.0 W of power. Calculate the energy it needs to provide light for 24 h. Answer in joules. **T/I** [ans:  $6.0 \times 10^5$  J]
2. Convert your answer in Question 1 to kilowatt hours. **T/I** [ans: 0.17 kWh]

$$\begin{aligned} 1. \quad P &= 7.0 \text{ W} \\ \Delta t &= 24 \text{ h} \\ &= 24 \cancel{\text{h}} \times \frac{60 \cancel{\text{min}}}{1 \cancel{\text{h}}} \times \frac{60 \text{ s}}{1 \cancel{\text{min}}} \\ &= 86\,400 \text{ s} \end{aligned}$$

$$P = \frac{\Delta E}{\Delta t}$$

$$\Delta E = P \Delta t$$

$$\begin{aligned} &= (7.0 \text{ W})(86\,400 \text{ s}) \\ &= 604\,800 \text{ J} \\ &= 6.0 \times 10^5 \text{ J} \end{aligned}$$

$$2) \quad 605\,000 \cancel{\text{J}} \times \frac{1 \text{ kWh}}{3\,600\,000 \cancel{\text{J}}}$$

$$\Delta E = 0.17 \text{ kWh}$$

## SOME WORK FOR YOU TO DO !

- Read 11.1 pgs. 504 – 507
- Pg. 507 #2-5
- Read 11.3 pgs. 510 - 513

